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A short synthesis of aspergillamide B. The marine natural product from *Aspergillus* sp.

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Abstract—A short route to the marine natural product aspergillamide **B** has been developed. The key step is the stereoselective formation of the *trans* indole–enamide pharmacophore by indole-assisted dehydration from the functionalized aminoalcohol intermediate. HPLC and ¹H NMR analyses show that aspergillamide **B** is more stable in the *cis* rotamer form. \bigcirc 2002 Elsevier Science Ltd. All rights reserved.

During the past few years, a number of bioactive natural products featuring an indole-3-ethenamide substructure have been isolated from various sources. Among them, terpeptine,¹ chondriamides² and coscinamides,³ as well as aspergillamides A (1a) and B (1b), are the most representative examples of this new class of compounds. The latters, cytotoxic peptides, have been structurally elucidated following their isolation in low yields from an *Aspergillus* sp. (Fig. 1).⁴

To date, a number of procedures allowing the formation of enamides have been developed.⁵ Interestingly, a library of aspergillamide analogues were recently prepared by Dömling and co-workers⁶ using the Ugi multicomponent reaction.⁷ Several of these analogues were found to be more active than the natural aspergillamides. However, as a major drawback, the Ugi multicomponent reaction⁷ gave in general a mixture of regio- and diastereoisomers,⁶ and, stereocontrolled processes are usually required to overcome this difficulty.⁸

Herein, we describe the first stereoselective synthesis of aspergillamide **B** taking advantage of a simple dehydration of a suitably functionalized aminoalcohol intermediate as outlined in Scheme 1.

To design this sequence, we thought that the electron rich indole unit might assist water elimination to provide the enamide moiety, in its thermodynamically stable *trans* configuration (Scheme 1). Following these observations, we have examined the dehydration process starting from the aminoalcohols **3a** and **3b** (Scheme 2). These derivatives were obtained in moderate yield



Figure 1. Structure of aspergillamides.

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Scheme 1. The proposed synthetic pathway allowing to the *trans*-enamide products.



Scheme 2. Synthesis of aminoalcohols 3a and 3b.

(34%) from aldehydes **2a** and **2b**, respectively, following: (i) Henry reaction⁹ using nitromethane aldol condensation and (ii) subsequent hydrogenolysis of the nitroalcohol intermediates. Interestingly, cyanosilylation of the same starting material (**2b**) using ZnI_2 as catalyst¹⁰ afforded quantitatively the cyanohydrine intermediate which was then reduced with LAH in THF into compound **3b** in 72% overall yield.¹¹

The coupling of aminoalcohol **3a** or **3b** with *N*-Boc,*N*-Me-phenylalanine¹² (HOBT/EDC) was achieved in quantitative yield affording compounds **4a** and **4b** (95–98%), respectively, as equimolar mixture of two diastereoisomers (Scheme 3). Removal of the Boc-protecting group of compound **4b** provided the aminoalcohol **5b**.



8b (cis rotamer: more stable)

Scheme 3. Synthesis of aspergillamide **B**. *Reagents and conditions*: (i) HOBT, EDC, DMF, *N*-Boc,*N*-Me-phenylalanine; (ii) dioxane–HCl; (iii) APTS, toluene, 60°C; (iv) *N*-Ac-leucine, HOBT, EDC, DMF, DIEA; (v) 6% Na–Hg, Na₂HPO₄, dioxane–methanol.

Interestingly, as expected, upon acidic treatment (APTS-toluene) 5b underwent a clean conversion into the single *trans*-enamide product **6b**, as confirmed by ¹H NMR [δ (H-enamide)=6.24 ppm and J=14.6 Hz].¹³ Alternatively, **6a** and **6b** were obtained from **4a** and 4b, respectively, using the same dehydration process which led to the Boc-protected enamides 7a and 7b, followed by Boc-deprotection (7b \rightarrow 6b, 51%). The coupling of **6b** and the commercially available *N*-acetylleucine under standard conditions yielded the protected aspergillamide **8b** in 78% yield. The benzenesulfonyl protecting group of compound 8b was cleanly removed by treatment with Na-Hg¹⁴ to give aspergillamide 1b in 72% yield.¹⁵ In the present case, this protecting group was found to be more resistant to the cleavage when using more conventional procedures.¹⁶ Finally, HPLC analysis¹⁷ of aspergillamide **1b** showed a mixture of *cis*and *trans*-amide rotational isomers (see Fig. 1) which equilibrated in favour of the stable cis rotamer. The cis conformation was evidenced by the high field signal ($\delta_{\rm H}$ at -0.16 ppm) of one of the two hydrogens of the leucine methylene group facing the phenyl ring (Phshielding effect, see ¹H NMR),¹⁵ in line with literature.⁴

In conclusion, we have realized the first stereoselective synthesis of aspergillamide **B**, in short linear steps. In vitro cytotoxicity assays of aspergillamide **B** and the intermediates **6b** and **8b** were evaluated against the KB cell line. These products were found to be active in a micromolar range [IC₅₀: 10 μ M for **1b**, 3 μ M for **6b** and 2.3 μ M for **8b**]. The application of this strategy to the synthesis of other enamide-natural products and analogues (in indole and non-indole series) together with the photoisomerization aspergillamide A \rightarrow aspergillamide **B** is in progress and will be given in due course.

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References

- Kagamizono, T.; Sakai, N.; Arai, K.; Kobinata, K.; Osada, H. *Tetrahedron Lett.* **1997**, *38*, 1223–1226.
- Davyt, D.; Entz, W.; Fernandez, R.; Mariezcurrena, R.; Mombru, A. W.; Saldana, J.; Dominguez, L.; Coll, J.; Manta, E. J. Nat. Prod. **1998**, *61*, 1560–1563.
- Bokesch, H. R.; Pannell, L. K.; McKee, T. C.; Boyd, M. R. *Tetrahedron Lett.* 2000, 41, 6305–6308.
- Toske, S. G.; Jensen, P. R.; Kauffman, C. A.; Fenical, W. Tetrahedron 1998, 54, 13459–13466.
- (a) Brettle, R.; Mosedale, A. J. J. Chem. Soc., Perkin Trans. 1 1988, 2185–2195; (b) Xiao, D.; East, S. P.; Joullié, M. M. Tetrahedron Lett. 1998, 39, 9631–9632; (c) Wu, Y.; Esser, L.; De Brabander, J. K. Angew. Chem., Int. Ed. 2000, 39, 4308–4310 and references cited therein; (d) Shen, R.; Porco, J. A., Jr. Org. Lett. 2000, 2, 1333– 1336; (e) Wang, X.; Porco, J. A., Jr. J. Org. Chem. 2001, 66, 8215–8221; (f) Fürstner, A.; Brehm, C.; Cancho-Grande, Y. Org. Lett. 2001, 3, 3955–3957; (g) Ribéreau, P.; Delamare, M.; Célanire, S.; Quéguiner, G. Tetrahedron Lett. 2001, 42, 3571–3573; (h) Lin, S.; Danishefsky, S. J. Angew. Chem., Int. Ed. 2002, 41, 512–515.
- Beck, B.; Hess, S.; Dömling, A. Bioorg. Med. Chem. Lett. 2000, 10, 1701–1705.
- (a) Ugi, I. Angew. Chem., Intl. Ed. Engl. 1982, 21, 810– 819; (b) Ugi, I. J. Prakt. Chem. 1997, 339, 499–516.
- For an example of asymmetric Ugi reaction, see: Linderman, R. J.; Binet, S.; Petrich, S. R. J. Org. Chem. 1999, 64, 336–337 and references cited therein.
- For a recent review, see: Luzzio, F. A. *Tetrahedron* 2001, 57, 915–945.
- Evans, D. A.; Caroll, G. L.; Truesdale, L. K. J. Org. Chem. 1974, 39, 914–917.

- 11. Compound **3a**: ¹H NMR (250 MHz, CDCl₃): δ 8.10 (d, J = 8.1 Hz, 1H), 7.50 (m, 2H), 7.20 (m, 2H), 4.90 (m, 1H), 3.00 (m, 2H), 2.60 (bs, 3H), 1.60 (s, 9H); ¹³C NMR (62.5 MHz, CDCl₃): δ 149.60, 135.70, 128.40, 124.40, 122.60, 122.50, 121.99, 119.50, 115.30, 83.60, 66.20, 47.33, 27.74. HRMS calcd for C₁₅H₂₀N₂O₃: 276.147. Found: 276.148. Compound **3b**: ¹H NMR (400 MHz, CDCl₃): δ 7.96 (d, J = 8.4 Hz, 1H), 7.86 (m, 2H), 7.71 (d, 1H, J = 8.0 Hz), 7.68 (s, 1H), 7.47–7.33 (m, 3H), 7.26 (t, 1H, J = 7.7 Hz), 7.14 (t, 1H, J = 7.7 Hz), 6.7 (br s, 3H), 5.12 (dd, 1H, J = 3.3 and 4.0 Hz), 3.46–3.28 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 137.69, 135.43, 134.26, 129.46, 128.49, 126.95, 125.44, 124.98, 123.82, 120.69, 119.69, 113.81, 71.59, 64.79. MS (ESI⁺): m/z 633 (2M+1), 317 (MH⁺), 299 (MH⁺-H₂O).
- N-Boc,N-Me-phenylalanine was prepared in quantitative yield from N-Boc-phenylalanine (NaH, CH₃I, DMF): [α]_D=-28.2° (*c* 1, EtOH), lit. -28.1° (*c* 1, EtOH). See: Omamoto, K.; Abe, H.; Kuromizu, K.; Izumiya, N. *Mem. Fac. Sci. Kyushu Univ. Ser. C* 1974, *9*, 131–138.
- 13. Compound **6b**: ¹H NMR (400 MHz, CDCl₃): δ 9.26 (d, 1H, J=11.0 Hz), 8.01 (d, 1H, J=8.1 Hz), 7.85 (m, 2H), 7.69 (d, 1H, J=7.7 Hz), 7.54–7.45 (m, 2H), 7.38–7.15 (m, 10H), 6.24 (d, 1H, J=14.6 Hz), 3.90 (bs, 1H), 3.39 (dd, 1H, J=4.4 and 7.0 Hz), 3.22 (dd, 1H, J=4.4 and 7.0 Hz), 2.82 (m, 1H), 2.32 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 171.03, 138.06, 137.00, 135.60, 133.93, 129.36, 129.18, 128.08, 127.21, 126.81, 125.16, 123.69, 123.37, 122.26, 120.46, 119.49, 113.87, 104.09, 65.78, 38.90, 35.38. MS (ESI⁺): m/z 482 (M+Na), 460 (M+1), 148.
- (a) Trost, B. M.; Arndt, H. C.; Strege, P. E.; Verhoeven, T. R. *Tetrahedron Lett.* **1976**, 3477–3478. For recent examples, see: (b) Sasaki, N. A.; Dockner, M.; Chiaroni, A.; Riche, C.; Potier, P. J. Org. Chem. **1997**, 62, 765–770; (c) Iradier, F.; Gomez Arrayas, R.; Carretero, J. C. Org. *Lett.* **2001**, *3*, 2957–2960.
- 15. Synthetic aspergillamide 1b: ¹H NMR (400 MHz, acetone-d₆): δ 10.25 (bs, 1H), 7.81–7.74 (m, 2H), 7.60–7.22 (m, 9H), 7.16–7.05 (m, 2H), 6.69 (d, 1H, J=14.6), 4.99 (dd, 1H, J=3.6 and 10.6 Hz), 4.60 (m, 1H), 3.31 (dd, 1H, J=3.6 and 14.3 Hz), 3.14 (dd, 1H, J=11.0 and 14.3 Hz), 2.88 (s, 3H), 2.01 (s, 3H), 1.60 (m, 1H), 1.19 (m, 1H), 0.80–0.68 (2d, 6H, J=6.8 Hz), -0.16 (m, 1H); ¹³C NMR (100 MHz, acetone-d₆): δ 172.86, 171.56, 166.18, 138.66, 137.38, 129.87, 129.28, 128.89, 128.32, 126.72, 125.61, 122.97, 121.78, 120.30, 119.48, 119.40, 112.79, 106.98, 62.54, 47.03, 38.01, 33.84, 28.62, 23.96, 22.67, 21.78, 19.75. MS (EI): *m/z* 474 (M⁺), 317.
- (a) Greene, T. W. Protective Groups in Organic Synthesis, 3rd ed.; Wiley-Interscience: New York, 1999; (b) Mg/ MeOH: Muratake, H.; Natsume, M. Heterocycles 1989, 29, 783–794.
- The reverse-phase HPLC analysis was carried out using a Waters Symmetry (4.6×250 mm) column and heptane– isopropanol (95/5, v/v) as eluant (1 mL/min).